Acta Crystallographica Section E

## **Structure Reports Online**

# benzylidene]hydrazine ISSN 1600-5368

## Xin Chen\* and Ming Yu

College of Sciences, Tianjin University of Science and Technology, Tianjin 300222, People's Republic of China

Correspondence e-mail: chen\_xin9999@163.com

#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(C-C) = 0.002 \text{ Å}$ R factor = 0.039 wR factor = 0.105 Data-to-parameter ratio = 15.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

In the title compound, C<sub>21</sub>H<sub>19</sub>N<sub>3</sub>O<sub>4</sub>, the central benzene ring makes dihedral angles of 75.46 (6) and 4.07 (6)° with the terminal benzene ring and the phenylhydrazine mean plane, respectively. The packing is stabilized by intermolecular N-H···O hydrogen bonds and weak non-classical intermolecular C-H···O hydrogen-bonding interactions.

(*E*)-1-(4-Nitrophenyl)-2-[4-(2-phenoxyethoxy)-

Received 21 September 2006 Accepted 25 September 2006

#### Comment

Metal complexes based on Schiff bases have attracted much attention because of their biological activity (Kahwa et al., 1986; Klayman et al., 1979). Consequently, a large number of Schiff base derivatives have been synthesized and employed to develop protein and enzyme mimics, such as models to mimic hydrolase in the hydrolysis of p-nitrophenyl picolinate (Li et al., 2005). 1-(4-Nitrophenyl)hydrazine forms a variety of Schiff bases with aldehydes, and the synthesis and crystal structures of some of them, such as (E)-1-(4-methoxy-3-propoxybenzylidene)-2-(4-nitrophenyl)hydrazine (Shi, 2005) and (E)-1-[4-(benzyloxy)benzylidene]-2-(4-nitrophenyl)hydrazine (Jun, 2005), have been reported.

In the present study we report the synthesis and molecular structure of the nitrophenylhydrazine Schiff base derivative (I) (Fig. 1).

The phenylhydrazine residue (C16–C21/N1/N2) is planar, with an r.m.s. deviation for fitted atoms of 0.0279 Å. This plane makes dihedral angles of 71.60 (6) and 4.07 (6)° with the terminal phenyl ring (C1-C6) and the central benzene fragment (C9-C15/O2), respectively. In addition, the dihedral angle between the terminal phenyl ring (C1-C6) and the central benzene fragment (C9-C15/O2) is 75.46 (6)°. The nitro group (O3/N3/O4) and its attached aromatic ring are not coplanar, with a dihedral angle of 3.70 (17)°. All bond lengths and angles are within normal ranges (Allen et al., 1987). The crystal packing is stabilized by intermolecular  $N{-}H{\cdots}O$ hydrogen bonds and weak C-H···O intermolecular hydrogen-bonding interactions that link molecules into onedimensional extended chains (Table 1 and Fig. 2).

© 2006 International Union of Crystallography All rights reserved

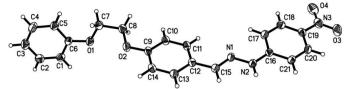


Figure 1
The molecular structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

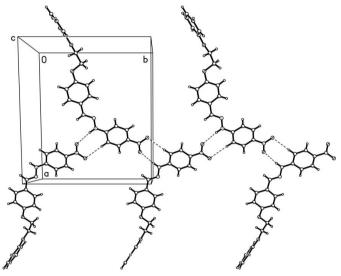


Figure 2
Packing diagram for (I), with hydrogen bonds drawn as dashed lines.

# **Experimental**

An anhydrous ethanol solution (50 ml) of 4-(2-phenoxyethoxy)-benzaldehyde (2.42 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 1-(4-nitrophenyl)hydrazine (1.53 g, 10 mmol) and the mixture stirred at 350 K for 5 h under nitrogen, giving a red precipitate. The product was isolated and recrystallized from acetonitrile. Red single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an acetonitrile solution.

# Crystal data

$C_{21}H_{19}N_3O_4$	Z = 8
$M_r = 377.39$	$D_x = 1.338 \text{ Mg m}^{-3}$
Monoclinic, C2/c	Mo $K\alpha$ radiation
a = 16.850 (5)  Å	$\mu = 0.09 \text{ mm}^{-1}$
b = 14.868 (4)  Å	T = 294 (2)  K
c = 15.190 (4)  Å	Block, red
$\beta = 100.120 (5)^{\circ}$	$0.34 \times 0.28 \times 0.22 \text{ mm}$
$V = 3746.3 (18) \text{ Å}^3$	

### Data collection

Bruker SMART APEX CCD areadetector diffractometer 3794 in  $\varphi$  and  $\omega$  scans 2184 re Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  $\theta_{\text{max}} = 0.948$ ,  $T_{\text{min}} = 0.948$ ,  $T_{\text{max}} = 0.980$ 

10330 measured reflections 3794 independent reflections 2184 reflections with  $I > 2\sigma(I)$   $R_{\rm int} = 0.037$   $\theta_{\rm max} = 26.4^{\circ}$ 

#### Refinement

Refinement on  $F^2$  w = 1/[e]  $R[F^2 > 2\sigma(F^2)] = 0.039$  + 0  $WR(F^2) = 0.105$  when S = 0.99  $(\Delta/\sigma)_{max} = 0.039$   $\Delta\rho_{max} = 0.039$   $\Delta\rho_{min} = 0.039$ 

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0478P)^2 \\ &+ 0.2798P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &= 0.001 \\ \Delta\rho_{\text{max}} &= 0.14 \text{ e Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.17 \text{ e Å}^{-3} \end{split}$$

**Table 1** Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$ \begin{array}{c} N2-H2\cdots O3^{i} \\ C21-H21\cdots O4^{i} \end{array} $	0.86	2.23	3.087 (2)	173
	0.93	2.56	3.485 (2)	176

Symmetry code: (i)  $-x + \frac{3}{2}$ ,  $y - \frac{1}{2}$ ,  $-z + \frac{3}{2}$ .

The H atoms were included in calculated positions and refined using a riding model, with C-H = 0.93 Å for  $Csp^2$  and 0.97 Å for methylene, N-H = 0.86 Å, and  $U_{iso}(H) = 1.2 U_{eq}(C,N)$ .

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

### References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans.* 2, pp. S1–19.

Bruker (1999). SMART (Version 5.0) and SAINT (Version 4.0) for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.

Jun, S. (2005). Acta Cryst. E61, o3895-o3896.

Kahwa, I. A., Selbin, J., Hsieh, T. C.-Y. & Laine, R. A. (1986). *Inorg. Chim. Acta*, 118, 179–185.

Klayman, D. L., Bartosevich, J. F., Griffin, T. S., Mason, C. J. & Scovill, J. P. (1979). J. Med. Chem. 22, 855–862.

Li, J.-Z., Xu, B., Li, S.-X., Zeng, W. & Qin, S.-Y. (2005). *Transition Met. Chem.* **30**, 669–676.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Version 5.10 for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.

Shi, J. (2005). Acta Cryst. E61, o4091-o4092.